PATENT COOPERATION TREATY

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INTERNATIONAL PRELIMINARY REPORT ON PATENTABI

(Chapter II of the Patent Cooperation Treaty)

(PCT Article 36 and Rule 70)

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Applicant's or agent's file reference PPD70156WO	FOR FURTHER ACTIO	R ACTION See Form PCT/IPEA/416					
International application No. PCT/GB2004/003497	International filing date (day/n 16.08.2004	nonth/year)	Priority date (day/month/) 30.09.2003	vear)			
International Patent Classification (IPC) or r C07C51/367	national classification and IPC		•				
Applicant		· .	· · · · · · · · · · · · · · · · · · ·				
SYNGENTA LIMITED et al.		: 					
This report is the international pro- Authority under Article 35 and tra	1. This report is the international preliminary examination report, established by this International Preliminary Examining Authority under Article 35 and transmitted to the applicant according to Article 36.						
2. This REPORT consists of a total	of 5 sheets, including this co	over sheet.	;	;			
3. This report is also accompanied	by ANNEXES, comprising:	1					
a. 🛭 sent to the applicant and	to the International Bureau) a	total of 2 sheets,	as follows:				
and/or sheets contain							
☐ sheets which superse	sheets which supersede earlier sheets, but which this Authority considers contain an amendment that goes beyond the disclosure in the International application as filed, as indicated in item 4 of Box No. I and the						
h ☐ /sent to the International i	b \Box (sent to the International Bureau only) a total of (indicate type and number of electronic carrier(s)), containing a						
sequence listing and/or ta	bles related thereto, in comp Listing (see Section 802 of	uter readable form o	only, as indicated in the	Supplemental			
Box Relating to Sequence	Elsting (see Section 602 of	uio Administrativo ii	iotraditorioy.	•			
4. This report contains indications r	elating to the following items:			•			
☐ Box No. I Basis of the op	inion	•	. •	$(x_{i_1}, \dots, x_{i_{m-1}}, \dots, x_{i_{m-1}})$			
☐ Box No. II Priority		•	•				
☐ Box No. III Non-establishr	nent of opinion with regard to	novelty, inventive s	tep and industrial applic	ability			
☐ Box No. IV Lack of unity o							
☐ Box No. V Reasoned stat applicability; ci	Box No. V Reasoned statement under Article 35(2) with regard to novelty, inventive step or industrial applicability; citations and explanations supporting such statement						
☐ Box No. VI Certain docum	ents cited		: i				
E .	in the international application						
☐ Box No. VIII *Certain observ	ations on the international ap	plication	•				
Date of submission of the demand	Dat	te of completion of this	report				
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Name and mailing address of the internation	nal Aut	thorized Officer		mas Patens.			
preliminary examining authority:		•	• •	September 11 g			
European Patent Office D-80298 Munich		elmann,:M	•				
Tel. +49 89 2399 - 0 Tx: 523 Fax: +49 89 2399 - 4465		lephone No. +49 89 23	199-8335	The state of the s			
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International application No. PCT/GB2004/003497

INTERNATIONAL PRELIMINARY REPORT ON PATENTABILITY

	Вох	No. I	Basis of the repo	rt			•			
1.	With filed	With regard to the language, this report is based on the international application in the language in which it was filed, unless otherwise indicated under this item.								
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	1-7			filed with telefax	on 02.11.2005					
		a sequ	uence listing and/or	any related table(s) - see Suppleme	ntal Box Rel	ating t	o Seque	nce Listing	
3.		The a	mendments have re	sulted in the canc	ellation of:				•	
			description, pages						٠.	
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INTERNATIONAL PRELIMINARY REPORT ON PATENTABILITY

Box No. V Reasoned statement under Article 35(2) with regard to novelty, inventive step or industrial applicability; citations and explanations supporting such statement

1. Statement

Novelty (N) Yes: Claims 1-7

No: Claims

Inventive step (IS) Yes: Claims 1-7

No: Claims

Industrial applicability (IA) Yes: Claims 1-7

No: Claims

2. Citations and explanations (Rule 70.7):

see separate sheet

Box No. VIII Certain observations on the international application

The following observations on the clarity of the claims, description, and drawings or on the question whether the claims are fully supported by the description, are made:

see separate sheet

INTERNATIONAL PRELIMINARY REPORT ON PATENTABILITY (SEPARATE SHEET)

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Item V

Reasoned statement under Article 35(2) with regard to novelty, inventive step or industrial applicability; citations and explanations supporting such statement

D1 US 4 532 346

D2 US 4 625 053

D3 FR 763 374

D4 US 4 505 743

D5 XP 00 223 77 22

V.1 Novelty

D1 and D2 describe the preparation of optically active 2-(4-hydroxyphenoxy)-propionic acid or salts thereof by reacting hydroquinone with an optically active 2-halopropanoic acid in presence of a base (D1, col.3, lines 63-68 and claim 1; D2, claim 1). D3 relates to the improvement of the enantiomeric resolution on an industrial basis using sulfite as a reductive agent to avoid oxidative impurities (D3, page 4, lines 8-11). D4 and D5 disclose the further use of 2-(4-hydroxyphenoxy)-propionic acid in the preparation of herbicids, such as clodinafop.

Since none of the above-cited documents disclose the use of a "mild reducing agent", as defined in the present claim 1, in the production of optically pure (R) 2-(4-hydroxyphenoxy)-propionic, novelty could be recognized for the subject-matters of claims 1 to 7.

V.2 Inventive step

The closest related process is known from **D2**. In order to avoid side-reactions, such as disubstitution of hydroquinone, precipitation is performed during the reaction using a base (**D2**, col.2, line 50 and claim 1; col.4, lines 16-22). The present claimed process differs from the one of **D2** in that it requires a mild reducing agent and avoids precipitation. The technical problem presently posed is accordingly to provide a production process of optically pure (R) 2-(4-hydroxyphenoxy)-propionic on an industrial scale, i.e. simple purification procedure, reduced costs and high ee. The solution proposed is the use of a mild reducing agent. The claimed process has been shown to solve the technical problem posed by using sodium bisulfite as reducing agent. Former processes of production of (R)2-(4-hydroxyphenoxy)-propionic are suffering from the formation of coloured by-products due to the oxidation of

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hydroquinone and disubstituted hydroquinone. None of these processes disclose the use of a mild reducing agent as a potential solution to solve these problems. Different mild reducing agents can be used (claim 5), sulphite or bisulphite being more preferred (claim 6). It is known from the skilled person from his general knowledge that sulfite is a protecting agent against oxidation. This is supported by **D3** (**D3**, page 4, lines 8-11). This document is directed to the resolution of optical isomers in general and of adrenaline in particular (**D3**, page 1; page 2, lines 34-43). **D3** is accordingly not directed to an asymmetric reaction and is not confronted to the removal of coloured-impurities. The reducing properties of sodium sulphite are not determining in the choice of the reagents used therein. The purpose of sodium bisulphite in **D3** is to adjust the pH (**D3**, page 4, lines 42-49 or claim 4; claim 1). The skilled person confronted with the present technical problem of formation of oxidative impurities would therefore not be inclined to combine the teachings of **D2** and **D3** and come up to the proposed solution. An inventive step is accordingly acknowledged.

Item VIII

- 1. The expression "such as" used in claim 1 has no limiting effect on the scope of this claim. The features following this expression are accordingly entirely optional. Since these features are relevant, they have been formulated as part of dependent claims 4 and 5. They should accordingly have been removed from claim 1 in order to fullfill the requirement of conciseness of Article 6 PCT.
- 2. For consistency and clarity reasons, a reference to the definition provided in claim 1 for the "mild reducing agent" used in claim 7 should have been introduced therein, i.e. "mild reducing agent as defined in claim 1" (Article 6 PCT).

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CLAIMS

- 1. A process for producing R-2-(4-hydroxyphenoxy) propanoic acid or a salt thereof by reaction of hydroquinone or a salt thereof with a S-2-halopropanoic acid or a salt thereof in the presence of a mild reducing agent wherein the $-\varepsilon$ mild reducing agent is a neutral or a charged low oxidation state sulphur species, such as sulphur dioxide, a sulphite, a bisulphite, a hydrosulphite, a metabisulphite, a sulphenic acid, a sulphinic acid, for example formamidine sulphinic acid, or a low oxidation state phosphorous species such as a phosphite or hypophosphite, or hydrazine, a hydrazine derivative, or ascorbic acid.
- 2 A process according to claim 1 wherein the S-2-halopropanoic acid is S-2chloropropanoic acid.
- 3 A process according to claim 1 or claim 2 wherein the excess hydroquinone is recovered for recycle.
- 4 A process according to any preceding claim wherein the mild reducing agent is a neutral or a charged low oxidation state sulphur species, a low oxidation. state phosphorous species, hydrazine, a hydrazine derivative or ascorbic acid.
- A process according to claim 4 wherein the mild reducing agent is sulphur 5 dioxide, a sulphite, a bisulphite, a hydrosulphite, a metabisulphite, a sulphenic acid, a sulphinic acid, a phosphite, hypophosphite, hydrazine, a hydrazine derivative or ascorbic acid.
- 6 A process according to claim 5 wherein the mild reducing agent is an alkali metal sulphite or bisulphite.
- 7 A process for the manufacture of quizalofop-P-ethyl, haloxyfop-P-methyl, fluazifop-P-butyl, clodinafop, cyhalofop-butyl or fenoxaprop-P-ethyl by a) producing R-2-(4-hydroxyphenoxy)propanoic acid by reaction of hydroquinone or a salt thereof with S-2-halopropanoic acid or a salt thereof.

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in the presence of a mild reducing agent, b) reacting the R-2-(4-hydroxyphenoxy)propanoic acid with the appropriate halo-aryl or halo-heteroaryl moiety to give a R-2-((4-aryloxy or heteroaryloxy)phenoxy)propanoic acid and c) esterification of the acid from step b) to give quizalofop-P-ethyl, haloxyfop-P-methyl, fluazifop-P-butyl, clodinafop, cyhalofop-butyl or fenoxaprop-P-ethyl.